Acta Crystallographica Section E

Structure Reports Online

ISSN 1600-5368

S. M. Malathy Sony,^a
K. Palani,^a P. Charles,^a
M. N. Ponnuswamy,^a*
N. Sureshbabu,^b
P. C. Srinivasan^b and
M. Nethaji^c

^aDepartment of Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India, ^bDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India, and ^cDepartment of Inorganic and Physical Chemistry, Indian Institute of Science, Bangalore 560 012, India

Correspondence e-mail: mnpsy2004@yahoo.com

Key indicators

Single-crystal X-ray study T = 293 K Mean $\sigma(C-C) = 0.003$ Å R factor = 0.045 wR factor = 0.143 Data-to-parameter ratio = 16.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-Benzyl-2-methyl-1-phenylsulfonyl-1*H*-indole

In the title compound, $C_{22}H_{19}NO_2S$, the two phenyl rings adopt a *trans* conformation with respect to the indole moiety. The indole system deviates slightly from planarity. The aromatic ring of the phenylsulfonyl group is almost perpendicular to the indole ring, while the benzyl ring is oriented at an angle of 77.01 (1)°. A zigzag $C-H\cdots O$ intermolecular hydrogen bond along the *b* axis and $C-H\cdots \pi$ interactions stabilize the crystal packing.

Received 20 December 2004 Accepted 25 January 2005 Online 5 February 2005

Comment

Indole and its derivatives have antimicrobial (Gadaginamath & Patil, 1999), anti-inflammatory (Rodriguez *et al.*, 1985), antibacterial (Okabe & Adachi, 1998) and antidepressive activities. Sulfates, sulfones and sulfonamides exhibit insecticidal, germicidal, antimicrobial and antibacterial activities (De-Benedetti *et al.*, 1985; Krishnaiah *et al.*, 1995), and certain phenyl sulfones show fungicidal properties (Wolf, 1999). In view of the above-mentioned pharmacological activities, the crystal structure of the title compound, (I), has been determined.

$$SO_2$$
 N
 CH_3
 (I)

The two phenyl rings in (I) adopts a *trans* conformation with respect to the indole moiety. The angles around atom S10 show significant deviation from regular tetrahedral, the largest deviation being found for the O-S-O angle [119.73 (9)°]. This widening is presumably the result of the repulsive interactions between the short S=O bonds (Sankaranarayanan, *et al.*, 2001; Govindasamy *et al.*, 1999; Rodriguez *et al.*, 1985). The sum of the angles (353.9°) around N15 and the value of the out-of-plane distance [0.191 (1) Å] with respect to the plane of the bonded atoms shows a slight pyramidalization, as observed in related indoles (Yokum & Fronczek, 1997; Sankaranar-

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

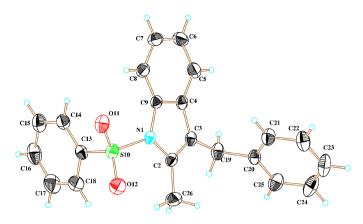


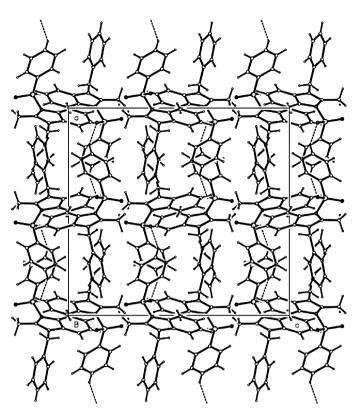
Figure 1A plot of the molecule of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

ayanan, et al., 2001; Beddoes et al., 1986). The indole system deviates slightly from planarity, as indicated by the dihedral angle between the pyrrole and benzene ring planes $[1.10 \text{ (3)}^{\circ}]$. In the phenylsulfonyl group, the phenyl ring is oriented almost perpendicular to the indole ring, at an angle of 83.93 (1)°. The benzyl ring is found to be oriented at an angle of 77.01 (1)° with respect to the indole ring. The methyl group at C2 is displaced significantly from the indole plane, by 0.104 (2) Å.

An intramolecular C−H···O hydrogen bond of pattern type S(6) between atoms C26 and O12 stabilizes the structure of (I) (Table 2). The molecules in the crystal structure form a zigzag chain along the b axis via intermolecular $C-H\cdots O$ hydrogen bonds of pattern type C(7) involving atoms C16 and O11 $(\frac{1}{2} - x, y - \frac{1}{2}, z)$. The pyrrole and benzene π systems of the centrosymmetric (-x, -y, 1-z) indole moieties are stacked over one another [dihedral angle 1.19 (6)°], the separation between their centroids being 4.662 (3) Å. The indole moiety interacts with the phenylsulfonyl ring of an inversion-related molecule at (-x, -y, 1-z) via a weak $C-H \cdot \cdot \cdot \pi$ interaction involving atom C5, the separation between H5 and the centroid of the C13–C18 ring being 3.09 Å. The phenylsulfonyl ring interacts with symmetry-related C20-C25 benzyl rings through weak $C-H\cdots\pi$ interactions involving atoms C15 and C17, the distances between atoms H15 and H17 and the centroids of the rings at (-x, -y, 1-z) and $(\frac{1}{2} + x, y, \frac{1}{2} - z)$ being 3.05 and 3.27 Å, respectively.

Experimental

2-Methyl-3-benzylindole (5.52 g, 25 mM) in dry tetrahydrofuran (THF) (50 ml) was slowly added to a stirred suspension of sodium hydride (1.2 g, 50 mM) in dry THF (10 ml) under nitrogen at room temperature. The reaction mixture was refluxed for 3 h under nitrogen and then cooled to 278 K, followed by the slow addition of phenylsulfonyl chloride (5 ml, 40 mM) in dry THF (25 ml). The reaction mixture was stirred at 273–278 K for 6 h, then treated with a saturated ammonium chloride solution (50 ml), and the layers were separated. The aqueous layer was extracted with dichloromethane



A packing diagram of the crystal structure of (I), viewed down the b axis. Dashed lines represent hydrogen bonds.

 $(4 \times 15 \text{ ml})$ and the combined organic layer was dried (Na₂SO₄) and concentrated *in vacuo* to give (I) as an oil, which was then crystallized from acetone.

Crystal data

$C_{22}H_{19}NO_2S$	Mo $K\alpha$ radiation		
$M_r = 361.44$	Cell parameters from 26 996		
Orthorhombic, Pbca	reflections		
a = 17.160 (9) Å	$\theta = 2.2 - 27.3^{\circ}$		
b = 11.680 (6) Å	$\mu = 0.19 \text{ mm}^{-1}$		
c = 18.274 (10) Å	T = 293 (2) K		
$V = 3663 (3) \text{ Å}^3$	Block, colourless		
Z = 8	$0.25 \times 0.22 \times 0.20 \text{ mm}$		
$D_{\rm x} = 1.311 \; {\rm Mg \; m^{-3}}$			

Data collection

Siemens SMART CCD area-	3235 reflections with $I > 2\sigma(I)$
detector diffractometer	$R_{\rm int} = 0.020$
ω scans	$\theta_{ m max} = 27.3^{\circ}$
Absorption correction: none	$h = -21 \rightarrow 21$
26 996 measured reflections	$k = -15 \rightarrow 15$
3882 independent reflections	$l = -22 \rightarrow 23$
•	

Refinement

-	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0872P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	+ 1.131 <i>P</i>]
$wR(F^2) = 0.143$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\text{max}} = 0.001$
3882 reflections	$\Delta \rho_{\text{max}} = 0.38 \text{ e Å}^{-3}$
235 parameters	$\Delta \rho_{\min} = -0.32 \text{ e Å}^{-3}$
H-atom parameters constrained	Extinction correction: none

Table 1 Selected geometric parameters (Å, °).

N1-C9	1.424 (2)	S10-O11	1.4271 (16)
N1-C2	1.432 (2)	S10-O12	1.4275 (17)
N1-S10	1.6650 (15)	S10-C13	1.754 (2)
C9-N1-C2	107.81 (13)	O12-S10-N1	107.20 (8)
C9-N1-S10	122.63 (11)	O11-S10-C13	108.89 (9)
C2-N1-S10	123.53 (12)	O12-S10-C13	108.73 (10)
O11-S10-O12	119.73 (9)	N1-S10-C13	105.57 (8)
O11-S10-N1	105.81 (8)		. ,
S10-N1-C2-C26	-30.4(2)	O11-S10-C13-C14	27.27 (17)
S10-N1-C9-C8	28.1 (2)	N1-S10-C13-C14	-85.95(16)
C9-N1-S10-O11	-39.92(14)	O12-S10-C13-C18	-21.35(19)
C2-N1-S10-O12	41.93 (16)	N1-S10-C13-C18	93.40 (18)
C9-N1-S10-C13	75.43 (14)	C2-C3-C19-C20	-86.5(2)
C2-N1-S10-C13	-73.88 (15)	C3-C19-C20-C21	-31.7(3)

Table 2 Hydrogen-bond geometry (Å, °).

Cg1 and Cg2 are the centroids of the C13-C18 and C20-C25 phenyl rings, respectively.

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
C26—H26A···O12	0.96	2.07	2.862 (3)	139
C16−H16···O11 ⁱ	0.93	2.56	3.399 (3)	151
$C5-H5\cdots Cg1^{ii}$	0.93	3.09	4.018 (4)	173
$C15-H15\cdots Cg2^{ii}$	0.93	3.05	3.701 (4)	129
C17 $-$ H17 \cdots Cg2 ⁱⁱⁱ	0.93	3.27	3.953 (4)	132

Symmetry codes: (i) $-x + \frac{1}{2}$, $y - \frac{1}{2}$, z; (ii) -x, -y, -z + 1; (iii) $x + \frac{1}{2}$, y, $-z + \frac{1}{2}$.

All H atoms were positioned geometrically (C-H = 0.93-0.98 Å) and allowed to ride on their parent atoms, with $U_{iso}(H) = 1.2$ or 1.5 times U_{eq} (parent atom).

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003), ORTEP3 (Farrugia, 1997) and ZORTEP (Zsolnai, 1998); software used to prepare material for publication: PLATON.

SMMS acknowledges the Council of Scientific and Industrial Research, India, for financial support. The authors also thank the Department of Science and Technology, India, for data collection on the CCD facility set up at the IIS, Bangalore, India, under the IRHPA-DST programme.

References

Beddoes, R. L., Dalton, L., Joule, J. A., Mills, O. S., Street, J. D. & Watt, C. I. F. (1986). J. Chem. Soc. Perkin Trans. 2, pp. 787-797.

De-Benedetti, P. G., Folli, U., Iarossi, D. & Frassineti, C. (1985). J. Chem. Soc. Perkin Trans. 2, pp. 1527-1532.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Gadaginamath, G. S. & Patil, S. A. (1999). Indian J. Chem. B, 38, 1070-1074. Govindasamy, L., Velmurugan, D., Shanmuga Sundara Raj, S. & Fun, H.-K. (1999). Acta Cryst. C55, 1315-1317.

Krishnaiah, M., Narayana Raju, K. V., Lu, I.-L., Chen, Y.-S. & Narasinga Rao, S. (1995). Acta Cryst. C51, 2429-2430.

Okabe, N. & Adachi, Y. (1998). Acta Cryst. C54, 386-387.

Rodriguez, J. G., Temprano, F., Esteban-Calderon, C., Martinez-Ripoll, M. & Garcia-Blanco, S. (1985). Tetrahedron, 41, 3813-3823.

Sankaranarayanan, R., Velmurugan, D., Shanmuga Sundara Raj, S., Fun, H.-K., Rao, S. N., Kannadasan, S. & Srinivasan, P. C. (2001). Acta Cryst. C57, 569-71.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Siemens (1996). SAINT (Version 4) and SMART (Version 5.6). Siemens Analytical X-Ray Systems, Inc., Madison, Wisconsin, USA.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.

Wolf, W. M. (1999). Acta Cryst. C55, 469-472.

Yokum, T. S. & Fronczek, F. R. (1997). Acta Cryst. C53, 362-363.

Zsolnai, L. (1998). ZORTEP. University of Heidelberg, Germany.